



The ligand-structure-selective binding of oligonucleotide by cobalt complexes

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ABSTRACT

The binding of two Co(III) complexes [Co(phen)₂(DPQ)]³⁺ and [Co(phen)₂(HPIP)]Cl₃ [HPIP = 2-(2-hydroxyphenyl)imidazo[4,5-f][1,10]phenanthroline, DPQ = dipyrido[3,2-f:2',3'-h]quinoxaline] to the normal base-paired decanucleotide d(CCTAATTAGG)₂ was studied by 2D NMR. The results indicate that the width of intercalating ligand has a large effect on the selectivity of binding site. For [Co(phen)₂(HPIP)]Cl₃, the complex binds the decanucleotide at C₂T₃:G₉A₈ and A₄A₅:T₇T₆ by intercalation from the minor groove, while [Co(phen)₂(DPQ)]³⁺ intercalates into T₃A₄:T₇A₈ region from the minor groove. The conclusion was further proved by molecular modeling.

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In the past decade, many octahedral Co(III) polypyridyl-type complexes have been synthesized as probes for DNA secondary structure, and their DNA-binding properties have been investigated mostly by UV–vis absorption spectra, emission spectra, viscosity measurements, circular dichroism spectra and gel electrophoresis experiments, etc. [1–12]. However, these methods are based on the properties of whole DNA molecules and could not provide the information about binding site. Two-dimensional NMR is powerful tool to detect DNA interactions and locate accurate binding sites, such as major/minor groove binding mode and base pairs involved in the complex intercalation [13–15].

In our previous work, we initially reported the binding of [Co(phen)₂(DPQ)]³⁺ to hexanucleotide d(GTCGAC)₂ by 2D NMR and molecular simulations [16]. However, as DNA model, hexamer is too short to form an integrated duplex. Herein, as an extension of our work, we studied the binding of the complex [Co(phen)₂(DPQ)]³⁺ and [Co(phen)₂(HPIP)]³⁺ to the decanucleotide d(CCTAATTAGG)₂ by 2D NMR (Fig. 1). The results suggest that the two complexes, with different width of the intercalating ligand, bind to the decanucleotide at different sites. [Co(phen)₂(DPQ)]Cl₃, with narrower ligand DPQ, intercalates into T₃A₄:T₇A₈ region from the minor groove, while [Co(phen)₂(HPIP)]³⁺, with wider ligand HPIP, intercalates into C₂T₃:G₉A₈ and A₄A₅:T₇T₆ region from the minor groove.

The ¹H NMR resonances of the free oligonucleotide acid have been assigned in the previous work [13]. The imino resonances spectrum indicates that only the terminal residues failed to form

stable base pairs (Supplementary material Fig. S1). It is therefore believed that the oligonucleotide is predominantly present as a stable duplex in the experimental condition.

Addition of [Co(phen)₂(HPIP)]³⁺ or [Co(phen)₂(DPQ)]³⁺ to d(CC-TAATTAGG)₂ induced large chemical shift changes for the ligand HPIP (especially H12, H13, H14, H15, H16) and DPQ (H11, H12, H13), while small shifts were observed for the ancillary ligand phen (Fig. 2). This behavior is consistent with preferential oligonucleotide binding of the ligand HPIP or DPQ by intercalation [13–15]. However, the chemical shift changes from the oligonucleotide are different for the two complexes binding. For [Co(phen)₂(HPIP)]³⁺ binding, a large upfield (~−0.771 ppm) of T₃Me in the major groove indicates the strong shield from the intercalating ligand, and reveals the binding site at T₃:A₈ region. While for [Co(phen)₂(DPQ)]³⁺ binding, significant upfield shifts were observed for the particular protons located in the minor groove in T₃A₄/T₇A₈ region, such as A₄H1' (−0.07 ppm), T₇H1' (−0.09 ppm), T₇H2'' (−0.08 ppm) and A₈H2'' (−0.10 ppm), which imply that the complex intercalates into T₃A₄:T₇A₈ from the minor groove.

NOESY spectra of d(CCTAATTAGG)₂ bound to the complexes (1:1) was recorded respectively at 20 °C with mixing time of 300 ms (Fig. 3 and Supplementary material Tables S3–S4). In addition to the expected intraduplex sequential NOEs from the oligonucleotide, a number of intermolecular NOE cross-peaks between complexes to d(CCTAATTAGG)₂ were observed (Fig. 3). For [Co(phen)₂(HPIP)]³⁺ binding, the intense NOE cross-peaks between phen and sugar are observed, including H3/8–G₉H1' (A), H3/8–T₃H1' (B), H5/6–T₃H1' (E), H5/6–A₈H1' (I), H4/7–A₈H1' (J), H4/7–T₃H1' (K). Because the sugar H1' protons are located in the decanucleotide minor groove, these NOEs indicate that the phen

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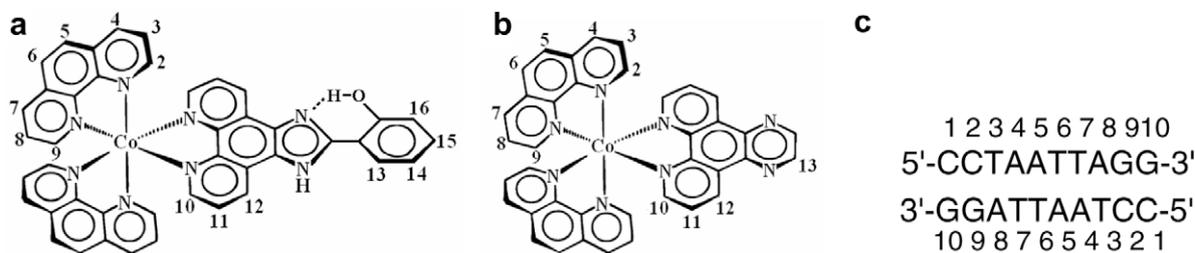


Fig. 1. The structure of $[\text{Co}(\text{phen})_2(\text{HPIP})]^{3+}$ (a) and $[\text{Co}(\text{phen})_2(\text{DPQ})]^{3+}$ (b) with atomic numbering; (c) the DNA sequence used in this work.

ligands are bound in the minor groove at the region $\text{C}_2\text{T}_3:\text{G}_9\text{A}_8$. This conclusion is further supported by the observation of NOE cross-peaks from phen and the decanucleotide $\text{H}_4'/\text{H}_5'/5''$ (protons located in the minor groove): $\text{H}_3/8-\text{T}_3\text{H}_5'/5''$ (C), $\text{H}_3/8-\text{C}_2\text{H}_4'$ (D), $\text{H}_5/6-\text{T}_3\text{H}_4'$ (G), $\text{H}_4/7-\text{T}_3\text{H}_4'$ (H). Therefore, the NOE data, combined with the upfield shift of T_3Me , indicate that the complex binds the decanucleotide by intercalation with the HPIP ligand selectively inserted from the minor groove and extended to the major groove between the stacked bases in $\text{C}_2\text{T}_3:\text{G}_9\text{A}_8$ region.

NOE cross-peaks from intercalating ligand HPIP to sugar protons were also observed; these included $\text{H}_{11}-\text{A}_5\text{H}_1'$ (M), $\text{H}_{11}-\text{A}_4\text{H}_1'$ (N), $\text{H}_{11}-\text{T}_6\text{Me}$ (O), and $\text{H}_{11}-\text{A}_5\text{H}_2''$ (P). Concomitantly, NOEs appeared from phen $\text{H}_5/6$ to minor groove proton $\text{T}_6\text{H}_1'$ (L) and $\text{A}_5\text{H}_4'$ (F). These NOE cross-peaks indicated that the complex binds the decanucleotide by intercalation between the stacked bases in $\text{A}_4\text{A}_5:\text{T}_7\text{T}_6$ region, with the HPIP ligand selectively inserted

from the minor groove. The disappearance of sequential intra-nucleotide NOEs $\text{A}_5\text{H}_2-\text{A}_4\text{H}_1'$ (Q) provides further evidence.

$[\text{Co}(\text{phen})_2(\text{DPQ})]^{3+}$ binding also induced a considerable number of intermolecular NOEs between the complex and $\text{d}(\text{CCTAATTAGG})_2$. Of note were the NOEs between phen to the protons located in the minor groove, such as $\text{H}_5/6-\text{A}_4\text{H}_1'$ (A), $\text{H}_5/6-\text{A}_8\text{H}_1'$ (B), $\text{H}_5/6-\text{T}_3\text{H}_1'$ (C), $\text{H}_5/6-\text{T}_3\text{H}_2''$ (F). It was not possible to unambiguously assign NOE cross-peaks between phen $\text{H}_5/6$ and the decanucleotide $\text{H}_4'/\text{H}_5'/5''$ (D) (protons located in the minor groove) due to the overlap of these resonances. Alternatively, the NOE cross-peak from H_{13} to major groove proton T_3Me (H) was also observed. These data suggest that $[\text{Co}(\text{phen})_2(\text{DPQ})]^{3+}$ intercalates into the $\text{T}_3\text{A}_4/\text{T}_7\text{A}_8$ region from the minor groove, with the DPQ ligand extending to the major groove. With the metal complex binding, some intra-nucleotide sequential NOEs missed or weakened, such as $\text{A}_4\text{H}_8-\text{T}_3\text{Me}$ (I) and $\text{A}_4\text{H}_8/\text{T}_3\text{H}_2'$ (J).

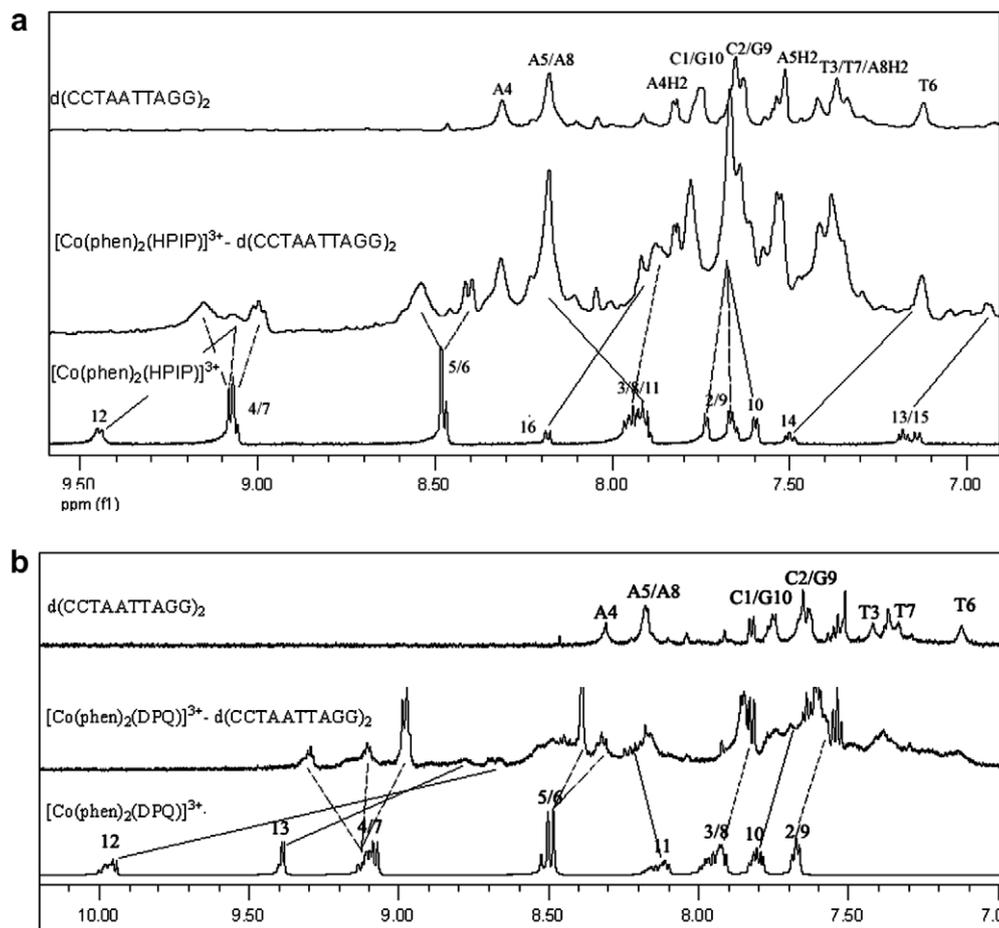


Fig. 2. 600 MHz ^1H NMR spectra of the complex-bound- $\text{d}(\text{CCTAATTAGG})_2$ at the ratio (R) of metal complex to duplex 1 (0.98 mM duplex) in the aromatic region.

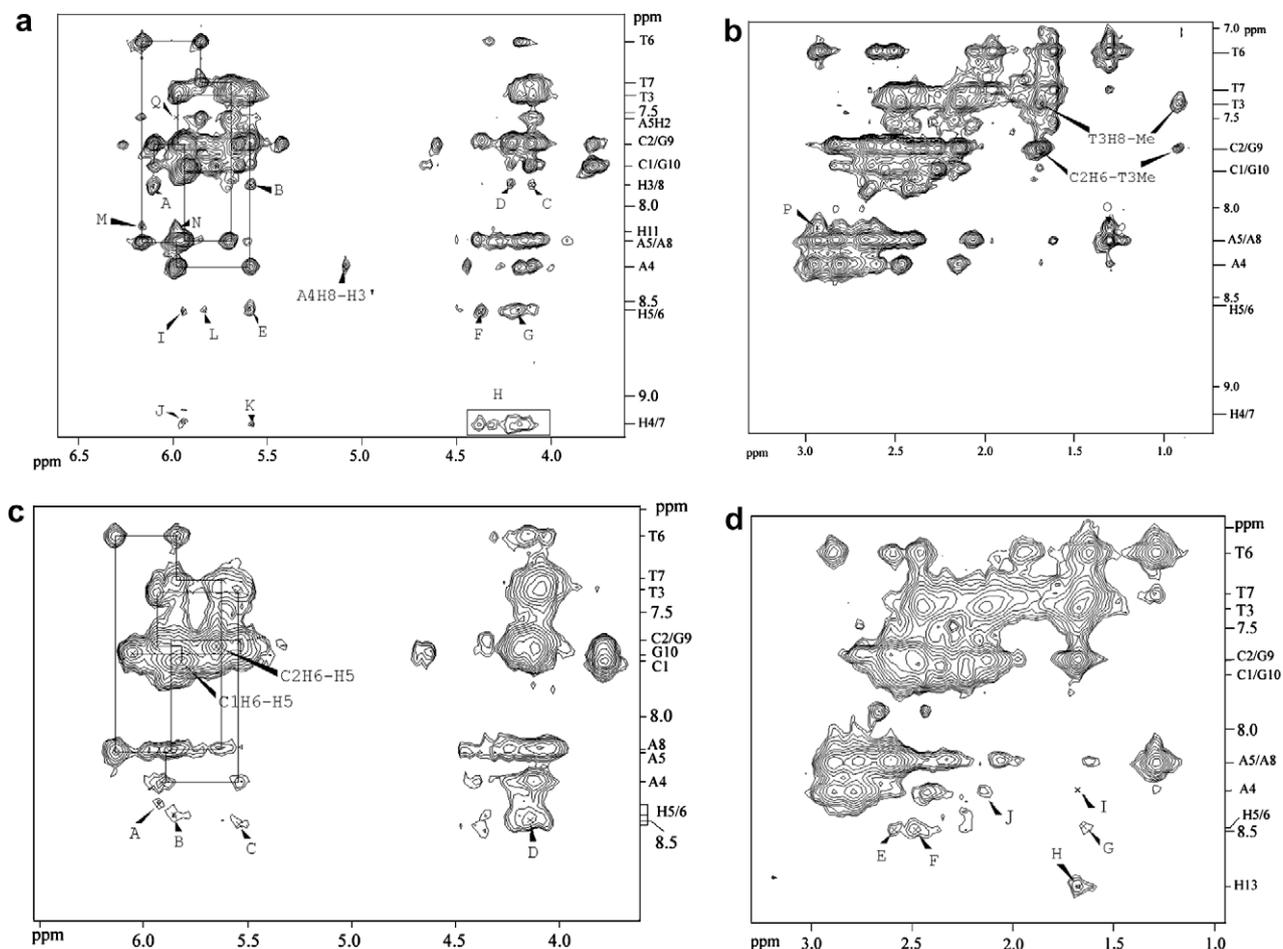


Fig. 3. Expansion of the NOESY spectrum (300 ms mixing time) of complex-bound-d(CCTAATTAGG)₂ at a metal complex-to-duplex ratio of 1. (a) The aromatic to the sugar H1' region of [Co(phen)₂(HPIP)]³⁺-bound-duplex; (b) the aromatic to the sugar H2'/H2'' region of [Co(phen)₂(HPIP)]³⁺-bound-duplex; (c) the aromatic to the sugar H1' region of [Co(phen)₂(DPQ)]³⁺-bound-duplex; (d) the aromatic to the sugar H2'/H2'' region of [Co(phen)₂(DPQ)]³⁺-bound-duplex. The NOEs from the cobalt complex resonances to the decanucleotide are labeled. X, Indicates the disappearance of the intrastrand sequential NOEs.

Further, to acquire the detailed binding information, a computational examination of interactions between the Co(III) complexes and the duplex was carried out by molecular modeling. The calculation results (Supplementary material Tables S5–S12) support the above conclusion.

In summary, although similar structure of the two complexes, the binding site with the same oligonucleotide is different. [Co(phen)₂HPIP]³⁺, with HPIP, intercalated into base pairs of a normal oligonucleotide duplex at two sites C₂T₃:G₉A₈ and A₄A₅:T₇T₆, while [Co(phen)₂(DPQ)]³⁺ intercalates into T₃A₄:T₇A₈ region from the minor groove. Maybe the difference in the binding site is derived from the width of the intercalating ligand. This result provides a theoretical basis for a rational design of a drug based on gene therapy.

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Appendix A. Supplementary material

The imino proton and sugar H1' ¹H NMR spectra of the free and complex-bound-duplex; Contour plots of the COSY spectrum of

free complex, TOCSY and NOESY spectra of d(CCTAATTAGG)₂-bound-complex; ¹H NMR Chemical Shift Data for free- and d(CCTAATTAGG)₂-bound-[Co(phen)₂(HPIP)]³⁺; The NOE cross peaks derived from the interaction between the complexes and d(CCTAATTAGG)₂; The data on the Molecular modeling. Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.inoche.2009.12.012.

References

- [1] M. Jiang, Y.T. Li, Z.Y. Wu, W.B. Peng, Q.L. Li, Polish Journal of Chemistry 83 (2009) 349–361.
- [2] F. Arjmand, M. Aziz, European Journal of Medicinal Chemistry 44 (2009) 834–844.
- [3] R.S. Kumar, S. Arunachalam, V.S. Periasamy, C.P. Preethy, A. Riyasdeen, M.A. Akbarsha, Australian Journal of Chemistry 62 (2009) 165–175.
- [4] (a) T.F. Miao, S.Y. Liao, L. Qian, K.C. Zheng, L.N. Ji, Biophysical Chemistry 140 (2009) 1–8; (b) B. Peng, H. Chao, B. Sun, H. Li, F. Gao, L.N. Ji, Journal of Inorganic Biochemistry 101 (2007) 404–411; (c) X.L. Wang, H. Chao, H. Li, X.L. Hong, Y.J. Liu, L.F. Tan, Journal of Inorganic Biochemistry 98 (2004) 1143–1150.
- [5] M.V. Alipazaga, R.G.M. Moreno, E. Linares, M.H.G. Medeiros, N. Coichev, Dalton Transaction 41 (2008) 5636–5644.
- [6] R.S. Kumar, S. Arunachalam, Biophysical Chemistry 136 (2008) 136–144.
- [7] E.K. Efthimiadou, A. Karaliota, G. Psomas, Bioorganic & Medicinal Chemistry Letters 18 (2008) 4033–4037.
- [8] R.S. Kumar, S. Arunachalam, V.S. Periasamy, C.P. Preethy, A. Riyasdeen, M.A. Akbarsha, Polyhedron 27 (2008) 1111–1120.
- [9] A.B. Olejniczak, P. Mucha, B. Gruner, Z.J. Lesnikowski, Organometallics 26 (2007) 3272–3274.

- [10] R. Indumathy, S. Radhika, M. Kanthimathi, T. Weyhermuller, B.U. Nair, *Journal of Inorganic Biochemistry* 101 (2007) 434–443.
- [11] S.J. Gomez, M.J. Prieto, B.M. Font, X. Solans, V. Moreno, *Inorganic Chemistry* 45 (2006) 10031–10033.
- [12] H.L. Chen, P. Yang, *Chinese Journal of Chemistry* 20 (2002) 1529–1535.
- [13] (a) H.L. Chen, P. Yang, C.X. Yuan, X.H. Pu, *European Journal of Inorganic Chemistry* (2005) 3141–3148;
(b) H.L. Chen, C.J. Dou, Y.B. Wu, X.L. Xi, W. Gao, P. Yang, *Inorganic Chemistry Communications* 12 (2009) 122–124;
(c) H.L. Chen, C.J. Dou, Y.B. Wu, H. Li, X.L. Xi, P. Yang, *Journal of Inorganic Biochemistry* 103 (2009) 827–832.
- [14] (a) J.A. Smith, J.G. Collins, B.T. Patterson, F.R. Keene, *Dalton Transactions* 9 (2004) 1277–1283;
(b) C.R. Brodie, J.G. Collins, J.R. Aldrich-Wright, *Dalton Transactions* 8 (2004) 1145–1152;
(c) J.R. Aldrich-Wright, J.G. Collins, F.R. Keene, Y. Tor, C. Brodie, W.A. Howard, F.M. Foley, B.T. Patterson, *Journal of Inorganic Biochemistry* 96 (2003) 90;
(d) J.G. Collins, A.D. Sleeman, J.R. Aldrich-Wright, I. Greguric, T.W. Hambley, *Inorganic Chemistry* 37 (1998) 3133–3141.
- [15] C.M. Dupureur, J.K. Barton, *Inorganic Chemistry* 36 (1997) 33–43.
- [16] Y.B. Wu, H.L. Chen, P. Yang, Z.H. Xiong, *Journal of Inorganic Biochemistry* 99 (2005) 1126–1134.